



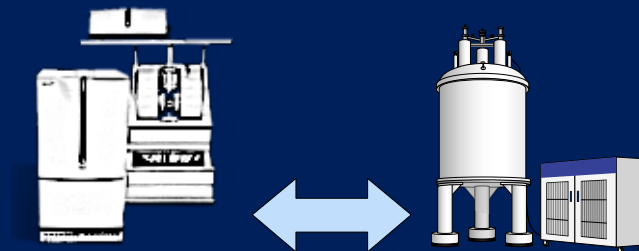
Electron Paramagnetic Resonance (EPR) at IOCB

Recent Case Studies & Future Instrumental Upgrades

Ján Tarábek

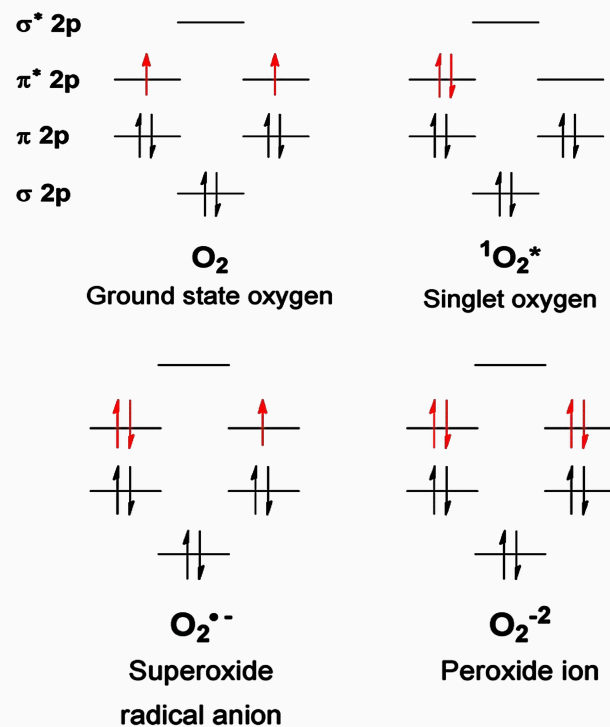
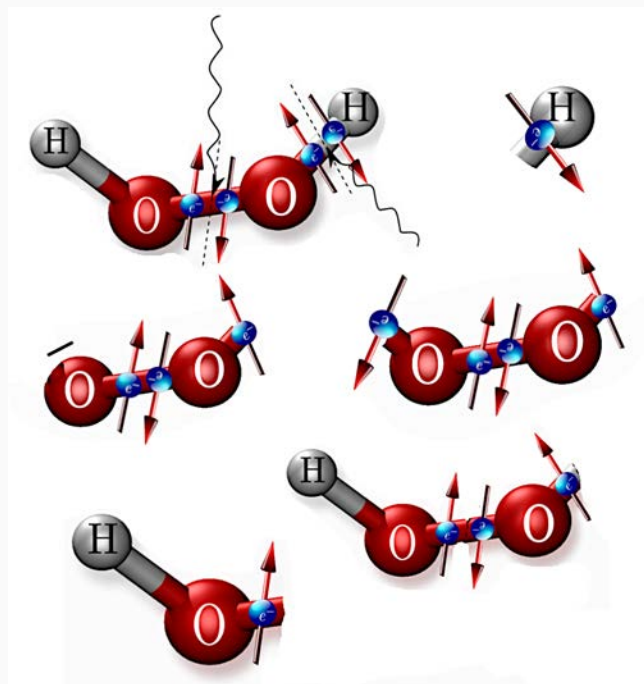


2021-09-23



EPR/NMR Essentials (Comparison)

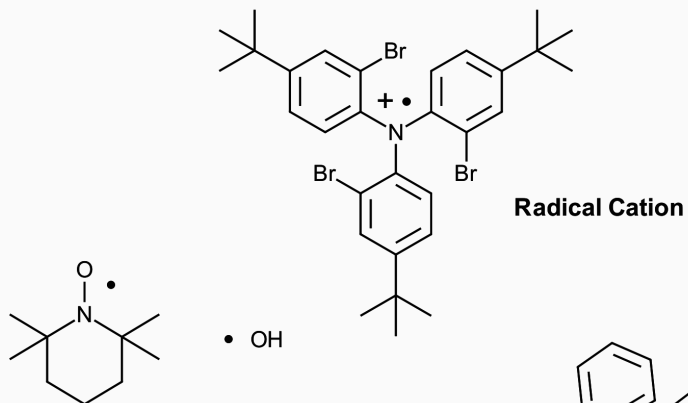
Simplified Scheme of Hydrogen Peroxide & ROS^a



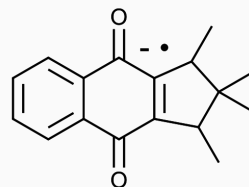
 ^a **ROS** $\equiv$ Reactive Oxygen Species

  K. Krumova & G. Cosa, [Overview of ROS in Singlet Oxygen: Applications in Biosciences and Nanosciences](#) **2016**, 1, pp. 1-21

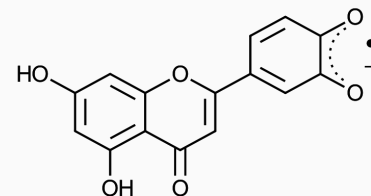
Paramagnetic Materials / Overview



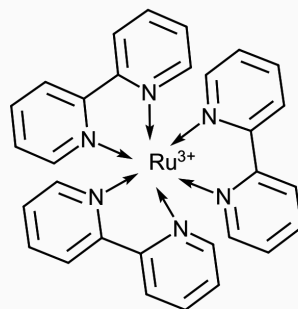
Neutral Radicals



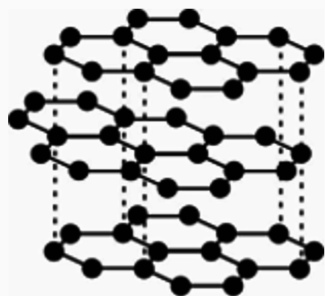
Radical Anions



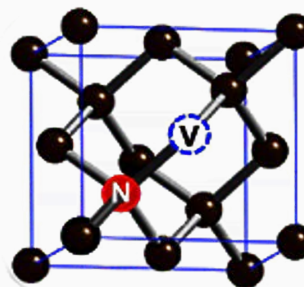
Biradical



Coordination Compound

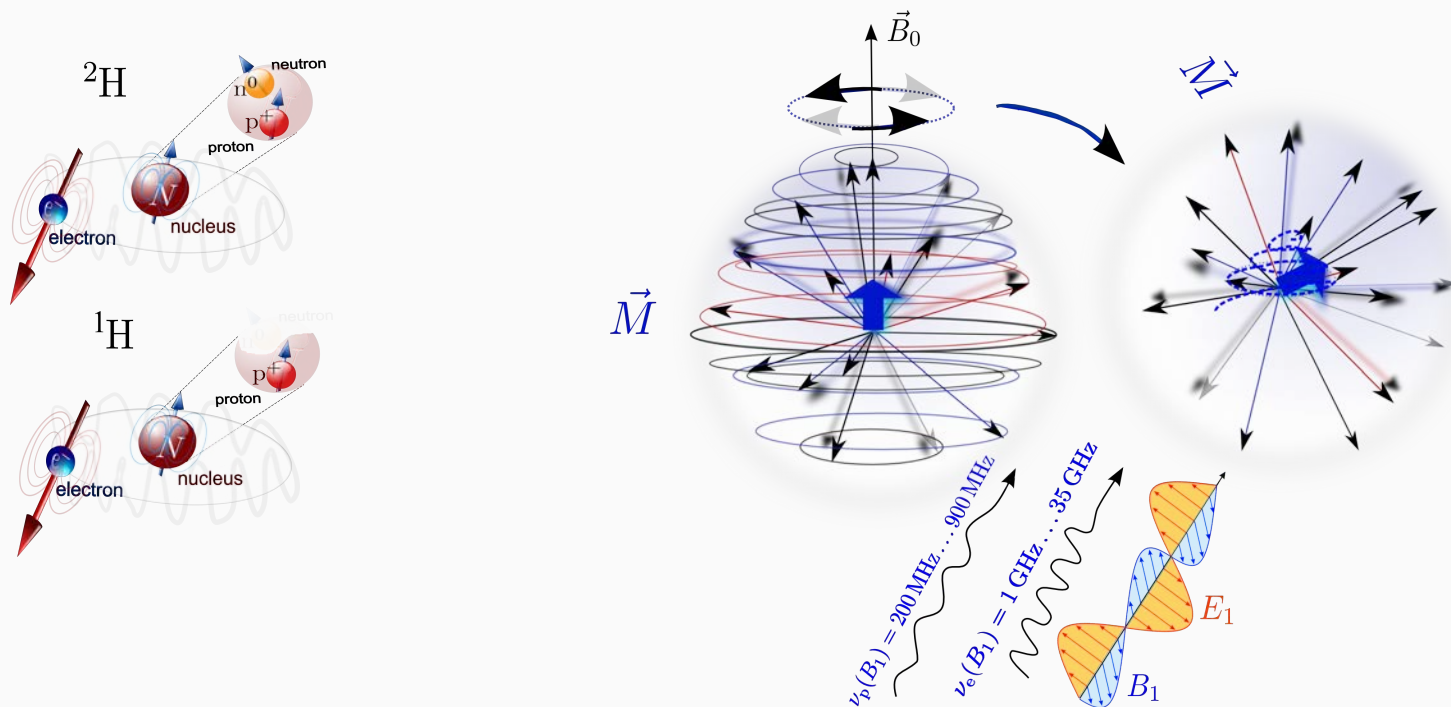


System of conducting electrons
(part of the **graphite** structure)



Part of the **nanodiamond**
structure with nitrogen
atom (N) & vacancy (V)

Common Features & Differences / Magnetism & Magnetic Resonance



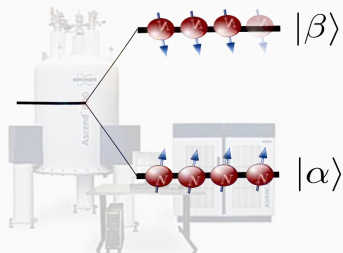
$\vec{B}_0 \Rightarrow$ external **magnetic flux density**/field

$\vec{M} \Rightarrow$ **magnetization**

ν_p (ν_e) $\Rightarrow$ matching **frequency for protons (electrons)**

Common Features & Differences / Energies & Population Difference

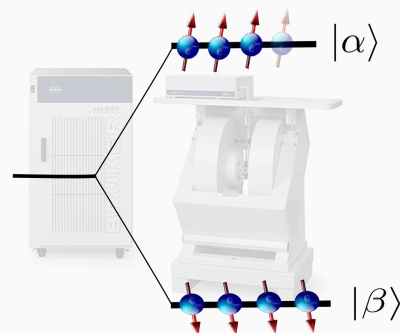
Proton (NMR)



$$\begin{aligned}\Delta\varepsilon_p &= h\nu_p \\ &= g_p \hbar \gamma_p B_0(\text{NMR})\end{aligned}$$

for $\nu_p = 600 \text{ MHz}$
 $B_0(\text{NMR}) = 14.0919 \text{ T}$
 $N_{|\beta\rangle}/N_{|\alpha\rangle} = 99.990 \% (T = 298 \text{ K})$

Electron (EPR)

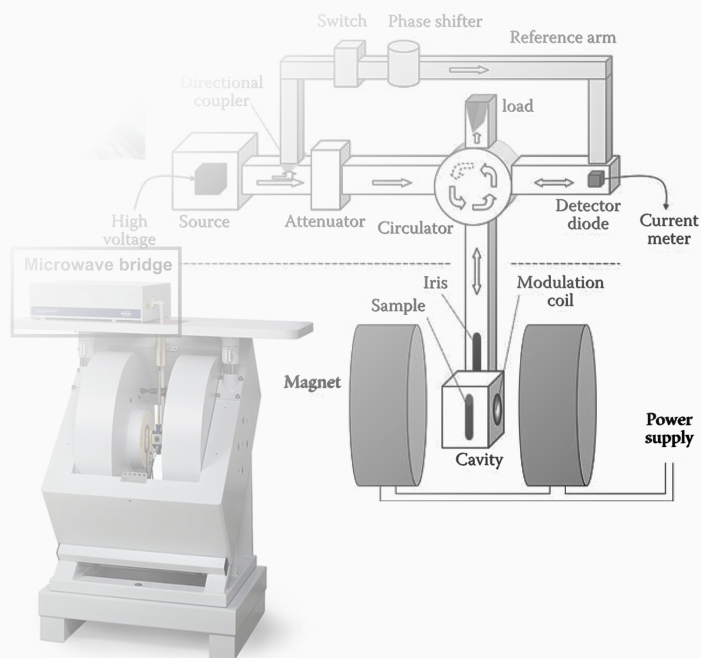


$$\begin{aligned}\Delta\varepsilon_e &= h\nu_e \\ &= g_e \hbar \gamma_e B_0(\text{EPR})\end{aligned}$$

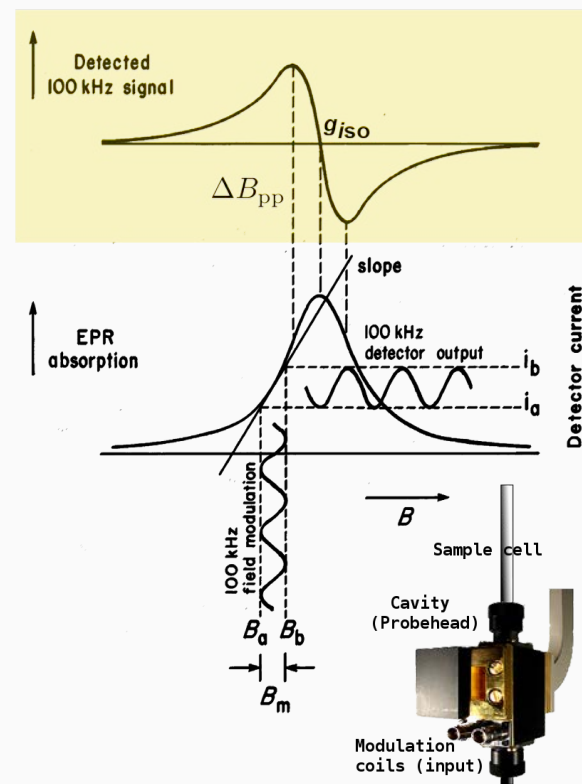
for $\nu_e = 9.8 \text{ GHz}$
 $B_0(\text{EPR}) = 0.3497 \text{ T}$
 $N_{|\alpha\rangle}/N_{|\beta\rangle} = 99.842 \% (T = 298 \text{ K})$

CW^b Spectrometer Detailed View & EPR Spectrum

Microwave Bridge and Magnet



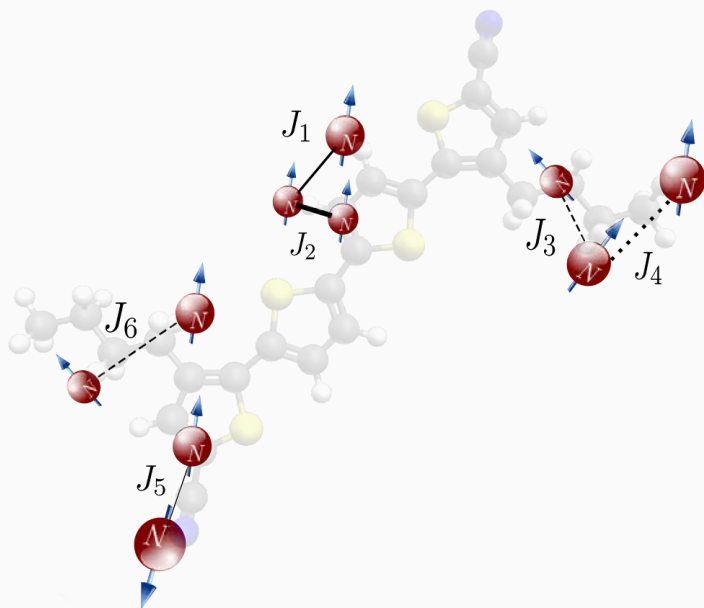
Origin of Signal/Spectrum



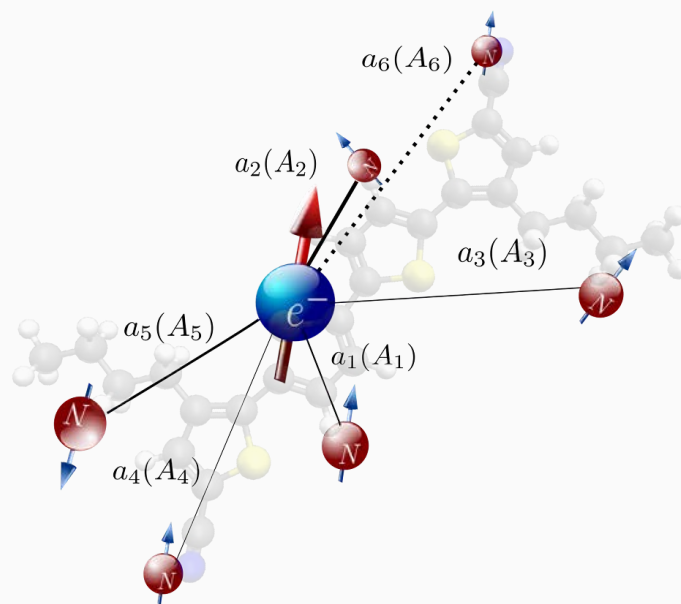
⚠ ^b CW $\equiv$ Continuous Wave $\Rightarrow \nu_e = \text{const.}$ $B_0 \neq \text{const.}$

Splitting of the NMR/EPR Spectra (**Nu-Nu/e-Nu** Interactions)

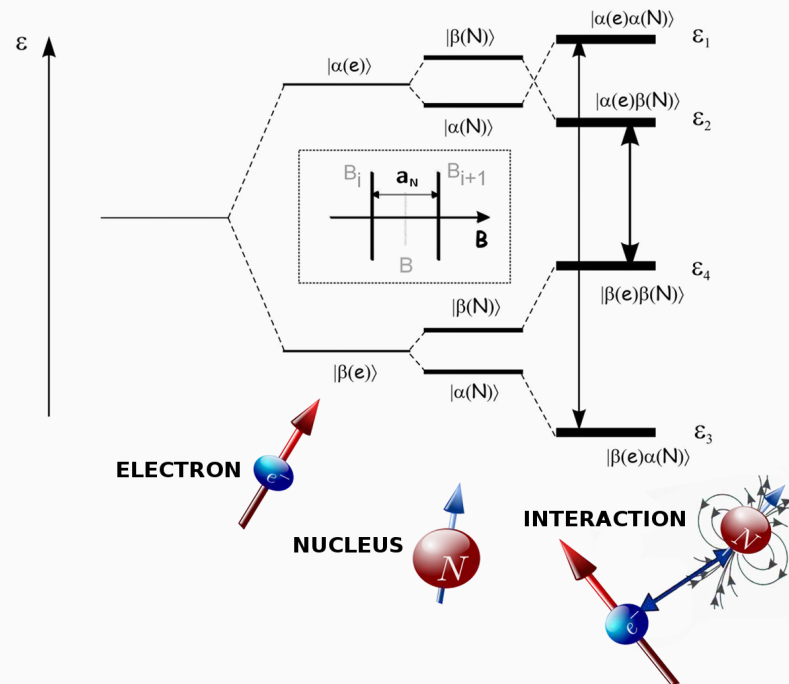
NMR



EPR



Example of $\mathbf{e}\text{-}^1\mathbf{H}$ Interaction (Energies ϵ)

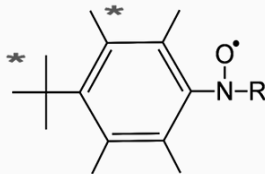


$$g \mu_B (B_{i+1} - B_i) = g \mu_B a_H \text{ (Splitting const.)} = A_H^\epsilon \text{ (Coupling const.)}$$

$$[a] = \text{mT (G)}$$

$$A^\epsilon \Rightarrow A^\epsilon/h \Rightarrow [A] = \text{MHz}$$

Example of Complex $e^{-}H^{-14}N$ Interaction/**Splitting**



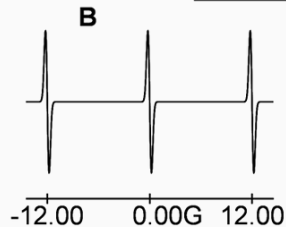
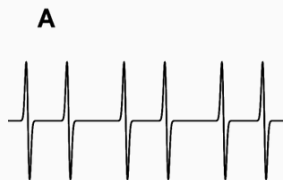
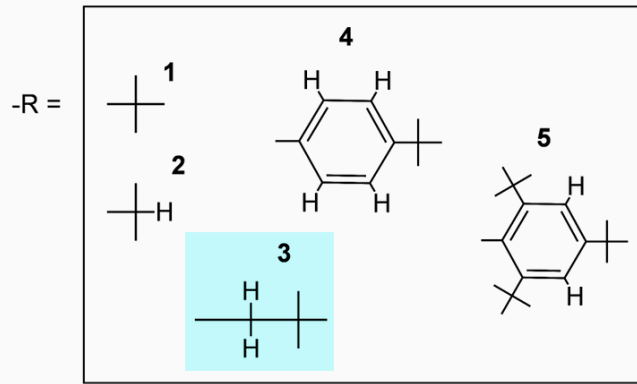
$$a_N (-NO-) = 12 \text{ G}$$

$$a_H (-CH-) = 5 \text{ G}$$

$$a_H (-CH_2-) = 10 \text{ G}$$

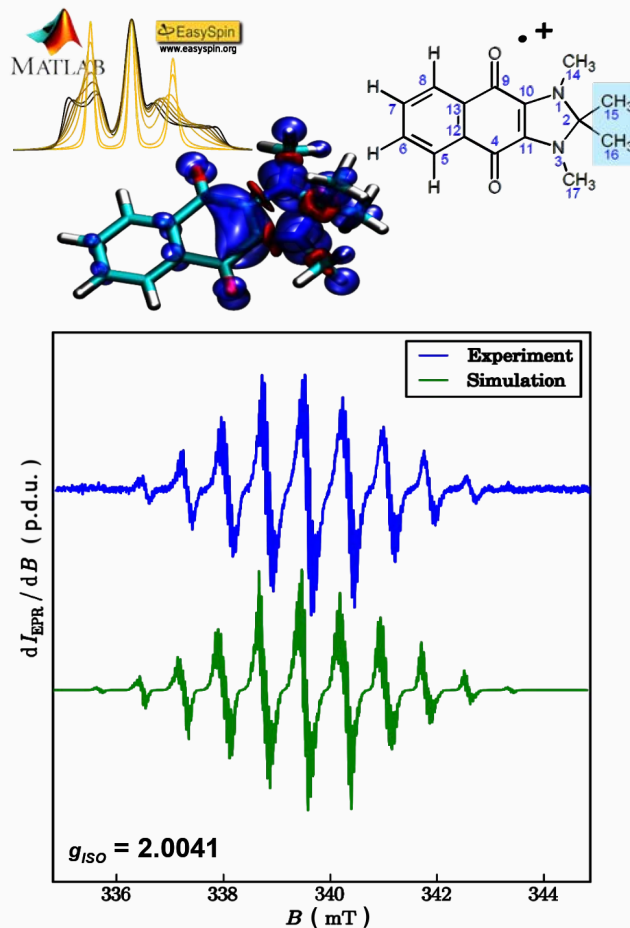
$$a_H^o (=CH-) = 3 \text{ G}$$

$$a_H^m (=CH-) = 1 \text{ G}$$

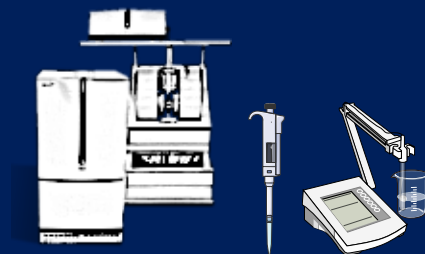


* Interaction/**Splitting** from *t* - **butyl** & -CH₃ can be **neglected**

c,d,e



^e see also: easyspin.org



Practical Aspects of EPR

EPR/NMR Spectrometers / Comparison



Tubes/Cells for Liquid/Solid Samples^{f,g} vs Temperature^h



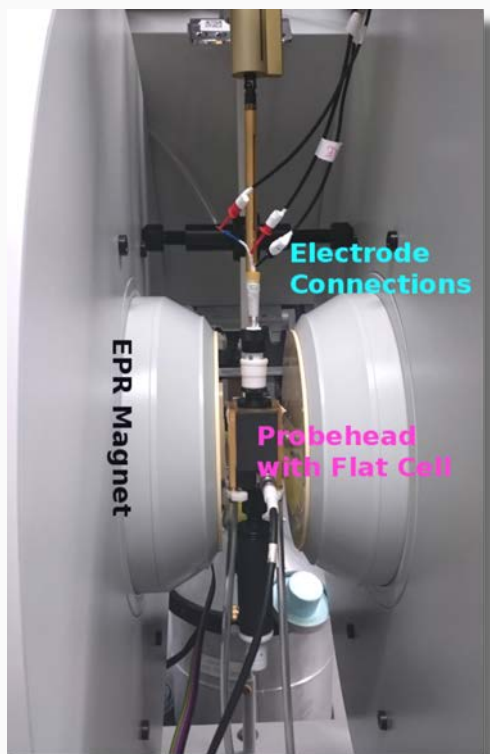
^f c region from 0.01 μM to 50 mM (V from 50 μl to 1 ml)

^g Powder m from 1 mg to 50 mg

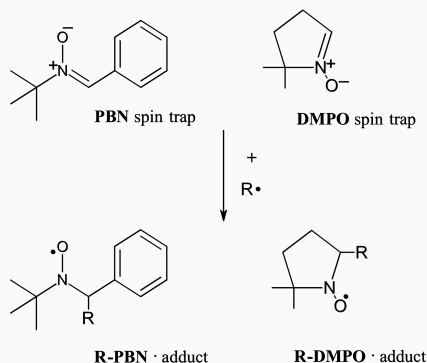
^h T from 80 K to 500 K

Coupled Methods/Generation of Paramagnetic Species

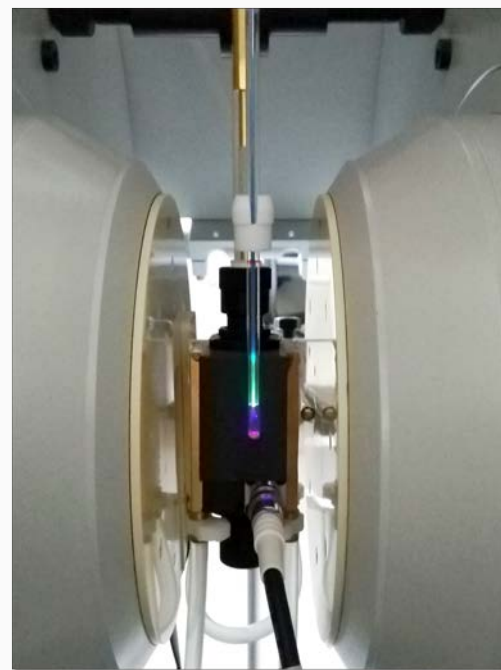
Electrochemistry

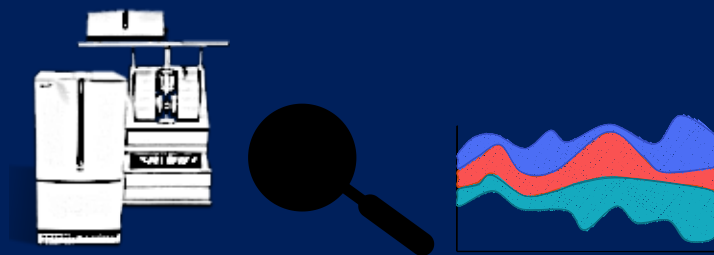


Spin Trapping



Photochemistry





Case Studies

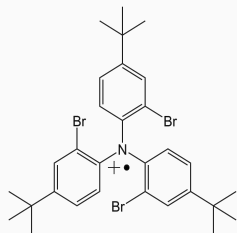


In situ EPR *Electrochemistry*



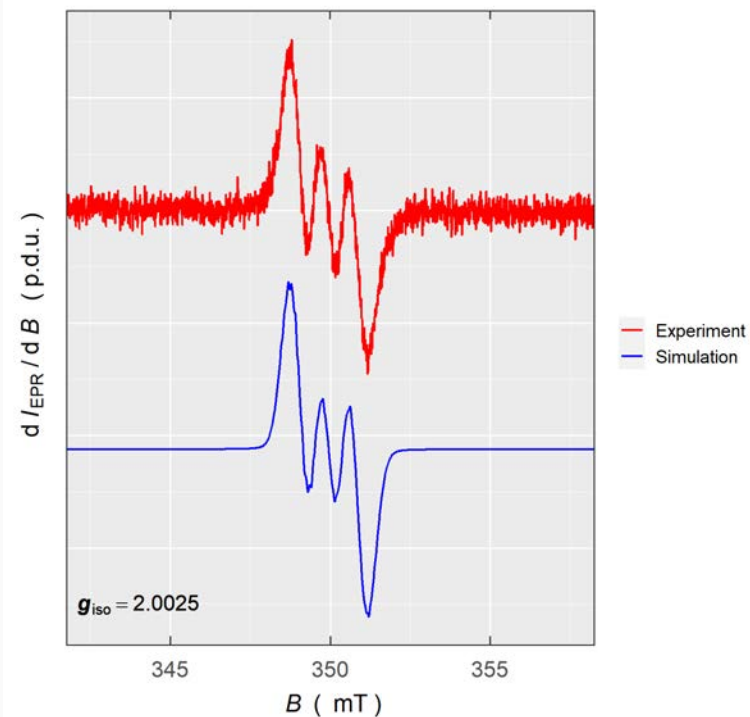
In situ EPR *Photochemistry* with Spin Trapping

Stability of the Electrochemically Generated **BC Radical Cation**



Hyperfine couplings (HFCCs, A) from simulation (Sim) + computed by DFT^j.

Nuclei	A^{Sim} (MHz)	A^{DFT} (MHz)
$1 \times {}^{14}\text{N}$	24.172	21.267
$3 \times {}^1\text{H}$	6.768	6.950
$3 \times {}^1\text{H}$	5.825	4.719
$3 \times {}^1\text{H}$	2.673	3.817
$3 \times {}^{79,81}\text{Br}$	1.820	2.036

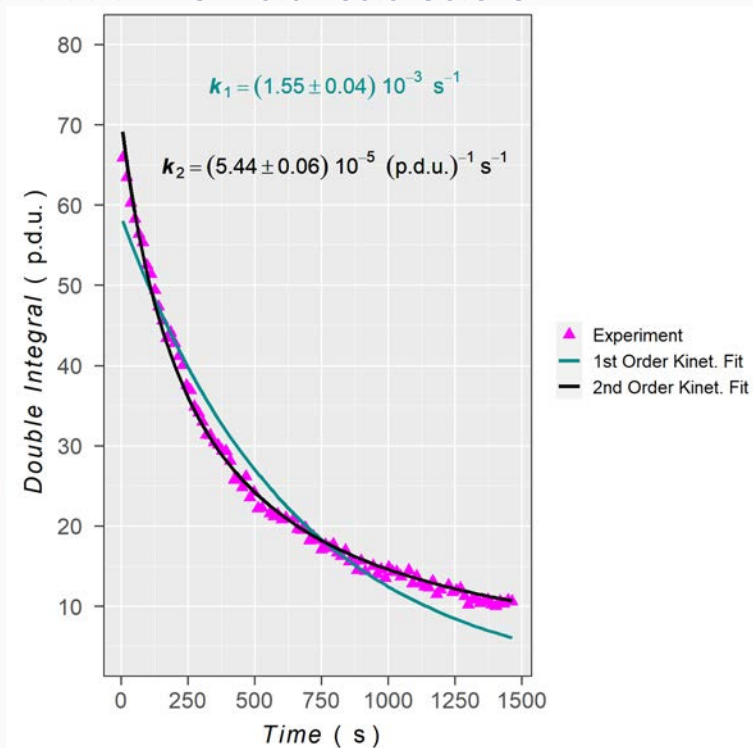
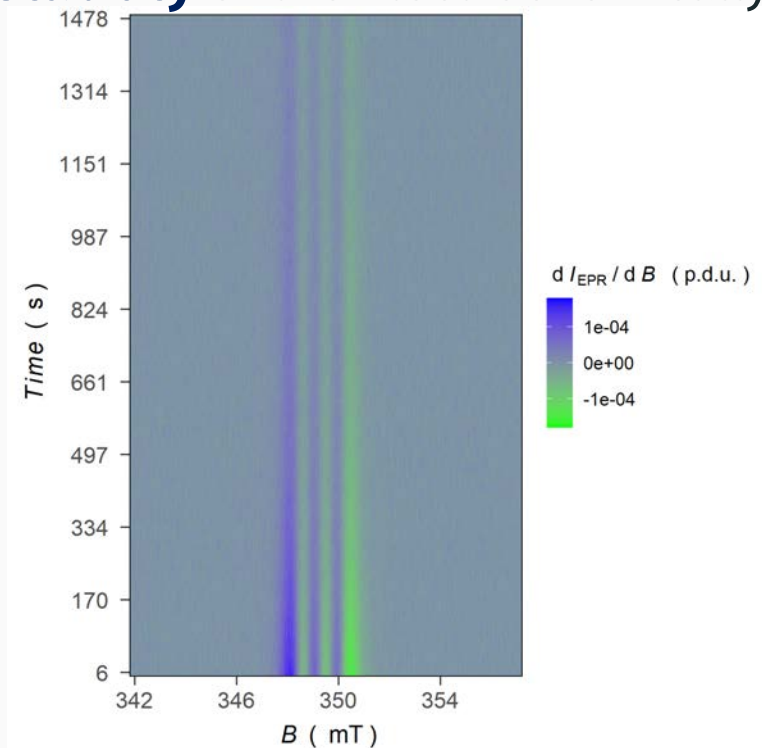


Comparison of experimental and simulated EPR spectra of "**Blues Cousin**"(BC), radical cation.

ⁱ Cooperation with the Group of [T. Slanina](#) (D. Hidasová, RP)

^j PBE0/EPR-II(C,H,N)/def2-TZVPD(Br)//B3LYP/6-31G(d,p)//C-PCM(CH₂Cl₂)

Stability of the Electrochemically Generated BC Radical Cation



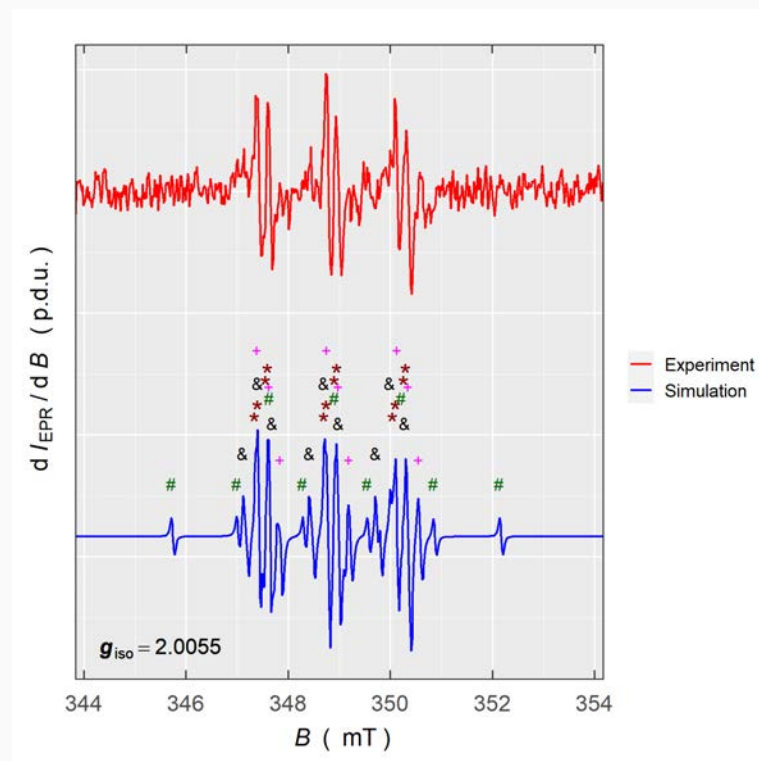
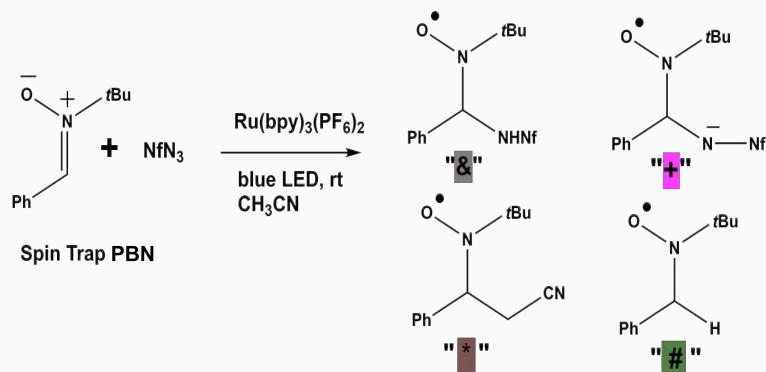
"Top-view" of EPR spectra recorded during kinetic stability experiment of the $\text{BC}^{\bullet+}$.

$\text{BC}^{\bullet+}$ decay^k just after the *in situ* electrochemical generation.

ⁱ Cooperation with the Group of [T. Slanina](#) ( D. Hidasová, RP)

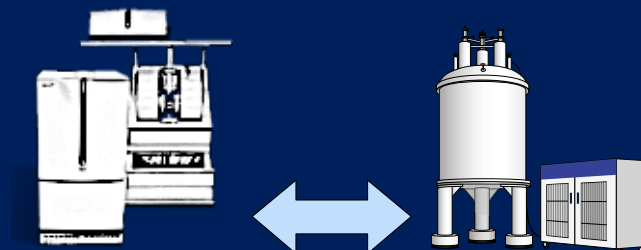
^k units => "p.d.u." $\equiv$ [procedure defined unit \(see IUPAC\)](#)

Detection of Transient Radicals by **Spin Trapping** upon **Continuous** Sample **Irradiation** (by 448 nm LED)^m



^l Cooperation with the Group of [U. Jahn](#) (👤 Ch. Pramthaisong, RP)

^m Additional **DFT** computations & **MS** & **Isotopic labeling/replacement** were required

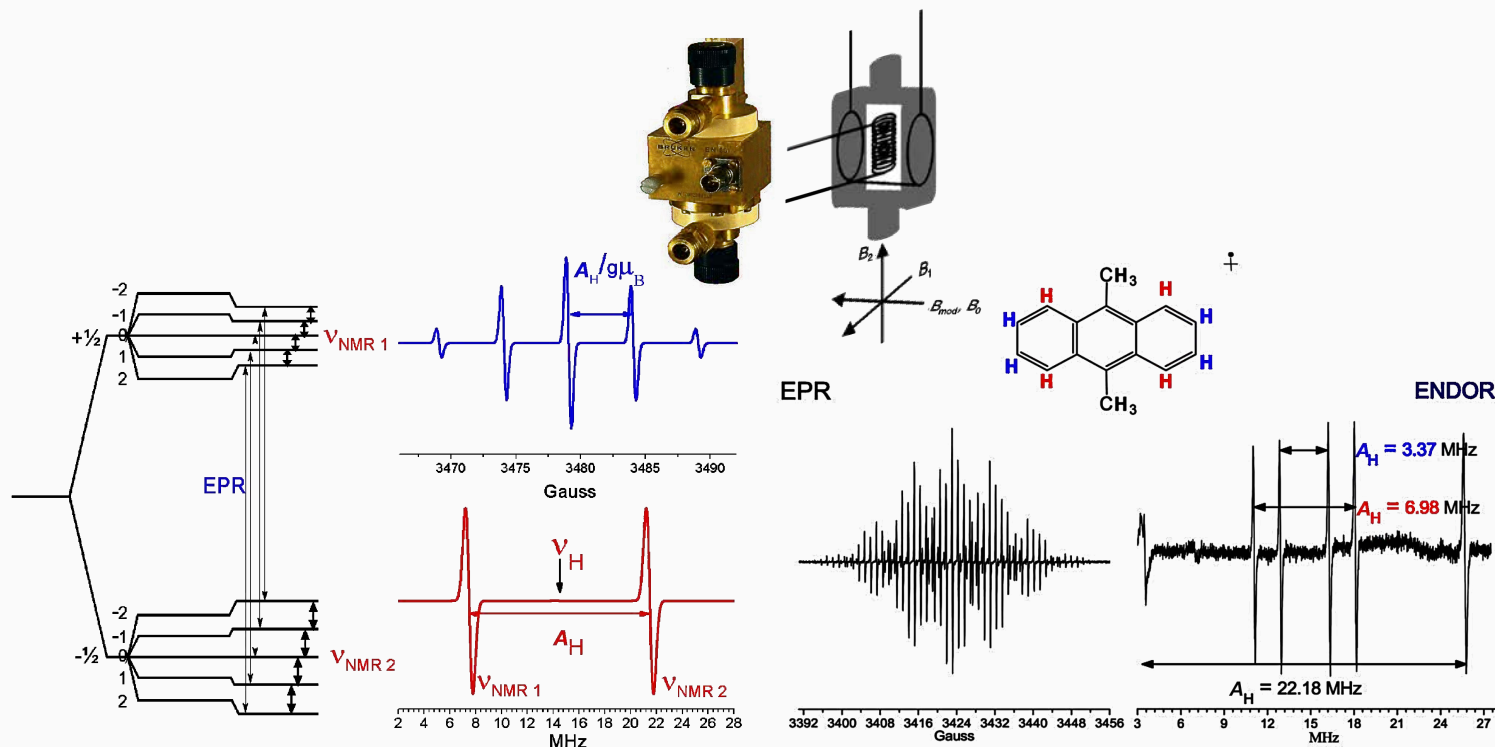


EPR Spectrometer Upgrade



Double Resonance

ENDOR Probehead & Spectra^o



ⁿ Electron Nuclear Double Resonance

^o doi [D. M. Murphy & R. D. Farley, Chem. Soc. Rev. 2006, 35, pp. 249-268](#)



Cooperation & Acknowledgements

- **NMR Team** 

- *P. Cígler* 

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- *I. Starý* 

- *T. Slanina* 

- *U. Jahn* 

- *J. Michl* 

- *M. Hocek* 

- *P. Bouř* 



TO BE CONTINUED ...